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A Question-Answering System for AP Chemistry: Assessing KR&R Technologies† Ken Barker¹, Vinay K. Chaudhri², Shaw Yi Chaw¹, Peter E. Clark³, James Fan¹, David Israel², Sunil Mishra², Bruce Porter¹, Pedro Romero⁴, Dan Tecuci¹, Peter Yeh¹ 1 U n iver sty of T x aA u, 2SR l l3B oei ng Pha t m W rks, 4 d-P p 1{ kb ar e, j chw f n po t ui zy }@ .s.x d 2{ ch aud ri, sel m}@ . o

A Question-Answering System for AP Chemistry: Assessing KR ...

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2002 AP Chemistry Exam (multiple choice only) 1999 AP Chemistry Exam; 1994 AP Chemistry Exam; These official exams come from before 2013 (when significant changes were made to the AP Chemistry curriculum), so they're formatted slightly differently from the current test. They have 75 multiple-choice questions (there are now 60) and six free ...

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course in chemistry with laboratory. The multiple-choice questions in Section I cover the breadth of the curriculum. Section II, the free-response part of the exam, includes three quantitative problems, one question on writing chemical reactions and predicting products, and two essays.

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CHEMISTRY - College Board

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AP Chemistry: Stoichiometry - Multiple Choice Answers 44. What number of moles of O_2 is needed to produce 14.2 grams of P_4O_{10} from P? (Molar Mass $P_4O_{10} = 284$) (A) 0.0500 mole (B) 0.0625 mole (C) 0.125 mole (D) 0.250 mole (E) 0.500 mole $4 P + 5 O_2 \rightarrow P_4O_{10}$ $\gg 14.2 \text{ g } P_4O_{10} \times \frac{1 \text{ mol } P_4O_{10}}{284 \text{ g } P_4O_{10}} = 0.0500 \text{ mol } P_4O_{10}$

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