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CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW SHORT

ANSWER Answer the following questions in the space provided.

1. Given the following equation:  $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$  a. What is the value of the coefficient x in this

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equation? 40.07 g/mol b. What is the molar mass of C<sub>3</sub>H<sub>4</sub>? 2 mol O<sub>2</sub>:1 mol H<sub>2</sub>O c. What is the mole ratio of O<sub>2</sub> to H

### **mc06se cFMsr i-vi - nebula.wsimg.com**

Preview this quiz on Quizizz. What is the maximum amount of product that can be formed between 2 or more reactants? Stoichiometry & Limiting Reactants - Chapter 9 Review. DRAFT. 9th - 12th grade. 259 times. Chemistry. 84% average accuracy. 2 years ago. jpfeiffer54. 0. Save. Edit. Edit. Stoichiometry & Limiting Reactants - Chapter 9 Review DRAFT ...

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5. Given the following unbalanced equation:  $\text{N}_2\text{O}(\text{g}) + \text{O}_2(\text{g}) \rightarrow \text{NO}_2(\text{g})$  a. Balance the equation. b. What is the mole ratio of NO<sub>2</sub> to O<sub>2</sub>? c. If 20.0 mol of NO<sub>2</sub> form, how many moles of O<sub>2</sub> must have been consumed? d. Twice as many moles of

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NO<sub>2</sub> form as moles of N<sub>2</sub>O are consumed. True or False? e.  
Twice as many grams of NO<sub>2</sub> form as grams of N<sub>2</sub>O are consumed. True or False?

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Stoichiometry Chapter 9 Test. Choose the best answer for each of the following questions. THINK through the questions. Pay attention to detail. Make sure all answers are in the correct amount of significant figures.

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Chapter 9 Stoichiometry Reading Quiz. Please read sections 9.1 A ,9.1 B, 9.2 A, and 9.2 B. Text only!! Do not do any practice problems from the book. I am trying to use numbers that can easily be done in your head without a calculator!! Try to use reasoning and estimating!!

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Introduction to Stoichiometry Much of our knowledge of chemistry is based on the careful quantitative analysis of substances involved in chemical reactions. Composition stoichiometry (which you studied in Chapter 3) deals with the mass relationships of elements in compounds.

## **CHAPTER 9 Stoichiometry - Quia**

Chapter 9 - Stoichiometry 9-1 Introduction to Stoichiometry  
Composition Stoichiometry - deals with mass relationships of elements in compounds  
Reaction Stoichiometry - Involves mass relationships between reactants and products in a chemical reaction  
I. Reaction Stoichiometry Problems A. Four problem Types, One Common Solution

## **Chapter 9 - Stoichiometry**

The explanation of why you can receive and get this chapter 9



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Reaction stoichiometry, the subject of this chapter, is based on chemical equations and the law of conservation of mass. All reaction stoichiometry ... The number of significant figures in the answer ... . Chapter 9 DO NOT EDIT--Changes must be made through "File info" ...

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